

# Synthesis of pyrophosphate coupled to the reverse energy-linked transhydrogenase reaction in *Rhodospirillum rubrum* chromatophores

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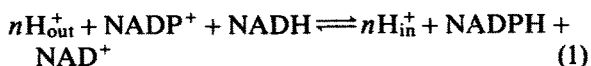
Received 11 February 1986

Chromatophores from the photosynthetic bacterium *Rhodospirillum rubrum* contain a membrane-bound transhydrogenase catalyzing the transfer of a hydride ion between NADH and NADP<sup>+</sup>. The reverse reaction, i.e. reduction of NAD<sup>+</sup> by NADPH, can furnish sufficient energy ( $\Delta\mu\text{H}^+$ ) to drive the phosphorylation of inorganic orthophosphate (P<sub>i</sub>) to pyrophosphate (PP<sub>i</sub>). The rate of PP<sub>i</sub> synthesis is 50 nmol PP<sub>i</sub> formed/min per  $\mu\text{mol}$  Bchl which is 5% of the rate of light-induced PP<sub>i</sub> synthesis. PP<sub>i</sub> synthesis is inhibited by both the H<sup>+</sup>-PPase inhibitor fluoride and the specific transhydrogenase inhibitor palmitoyl-CoA. The effects of both DCCD and uncouplers on the system provide additional evidence that the  $\Delta\mu\text{H}^+$  generated by the reverse transhydrogenase reaction drives PP<sub>i</sub> synthesis. The rate of PP<sub>i</sub> synthesis can be partially inhibited by the addition of NADP<sup>+</sup>, a substrate of the forward energy-consuming reaction. The  $\Delta\mu\text{H}^+$  generated can also be used to drive ATP synthesis by the H<sup>+</sup>-ATPase, but at a lower rate than the PP<sub>i</sub> synthesis.

(*Rhodospirillum rubrum*)      Energy-linked transhydrogenase      Pyrophosphate synthesis      ATP synthesis  
Bioluminescence

## 1. INTRODUCTION

The energy-linked nicotinamide nucleotide transhydrogenase (EC 1.6.1.1) catalyses the reduction of NADP<sup>+</sup> by NADH. The enzyme functions as a proton pump and translocates protons across the membrane according to reaction 1 (reviews [1,2]).



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**Abbreviations:** APS, adenosine 5'-phosphosulphate; Bchl, bacteriochlorophyll; BSA, bovine serum albumin; DAPP, diadenosine pentaphosphate (ApsA); DCCD, *N,N'*-dicyclohexylcarbodiimide; DTE, 1,4-dithioerythritol; FCCP, carbonyl cyanide *p*-trifluoromethoxyhydrazide; PCB<sup>-</sup>, phenyl dicarbaundecaborane; ANS, 8-anilinonaphthalene-1-sulphonic acid; H<sup>+</sup>-PPase, H<sup>+</sup>-pyrophosphatase; H<sup>+</sup>-ATPase, H<sup>+</sup>-ATP Synthase

Here, the subscripts out and in refer to H<sup>+</sup> locations in mitochondria or bacteria; the opposite is the case for submitochondrial particles and chromatophores. The first direct evidence for the existence of this enzyme, in submitochondrial particles, was provided by Danielson and Ernster [3.] Later, it was also shown to exist in bacteria [4-7] including the photosynthetic bacterium *Rhodospirillum rubrum* [8,9]. The energy coupling of the enzyme has been demonstrated by an energy-dependent 5-10-fold increase in the rate [3] and a 500-fold increase in the extent [10,11] of the reduction of NADP<sup>+</sup> by NADH. In mitochondria the transhydrogenase reaction can be driven by electron transport generated energy through any of the coupling sites of the respiratory chain, or by ATP hydrolysis [3,12]. In *R. rubrum* chromatophores, Keister and Yike [8,9] have shown that the reaction can be driven by ATP or PP<sub>i</sub> hydrolysis and by light.

Reversibility of the energy-linked transhydrogenase shown in reaction 1 has been demonstrated using ATP production [13], uptake of the lipophilic anion  $\text{PCB}^-$  [14–16] and by enhancement of ANS fluorescence [17]. The reaction has also been shown to induce the energy-dependent red shift of cytochrome *b* in intact mitochondria [18,19].

In addition to the  $\text{H}^+$ -ATPase, *R. rubrum* chromatophores also contain a membrane-bound  $\text{H}^+$ -PPase which catalyses both  $\text{PP}_i$  synthesis [20] and  $\text{PP}_i$  hydrolysis [21].

In these chromatophores  $\text{PP}_i$  synthesis can be achieved in both continuous [20] and intermittent light [22]. In view of the fact that the reverse transhydrogenase reaction can drive phosphorylation of ADP to ATP in submitochondrial particles [13], it seemed possible that it could also be used to drive  $\text{PP}_i$  synthesis by the  $\text{H}^+$ -PPase in the chromatophore membrane. We show here that this is the case, by monitoring continuous  $\text{PP}_i$  synthesis by a sensitive luciferase technique [23].

## 2. MATERIALS AND METHODS

Trizma base, glycylglycine, DTE, FCCP, oligomycin, antimycin A, valinomycin, nigericin, magnesium acetate, BSA, APS, DAPP, palmitoyl-CoA, NADPH, NADH,  $\text{NADP}^+$ ,  $\text{NAD}^+$  and ATP-sulphurylase (ATP:sulphate adenylyl transferase; EC 2.7.7.4) were purchased from Sigma (St. Louis, USA). Purified luciferase (EC 1.13.12.7), D-luciferin and L-luciferin were obtained from LKB Wallac (Turku, Finland). DCCD was obtained from Fluka (Buchs, Switzerland). Other materials were reagent grade and obtained from commercial sources.

### 2.1. Preparation of chromatophores

*R. rubrum* (strain S1) was grown anaerobically in the medium described in [24]. After 40 h of growth, cells were harvested and washed, and chromatophores prepared as in [25] with the exception that the cells were washed once in 0.2 M glycylglycine buffer, pH 7.4. Bchl was determined using an in vivo extinction coefficient of  $140 \text{ mM}^{-1} \cdot \text{cm}^{-1}$  at 880 nm reported by Clayton [26].

### 2.2. Continuous monitoring of $\text{PP}_i$ synthesis

Measurement of  $\text{PP}_i$  formation was performed as in [23]. All reactions were carried out at room

temperature. The standard assay volume of 1 ml contained the following components: 0.1 M glycylglycine, pH 7.75 (NaOH), 2 mM EDTA, 10 mM  $\text{Mg}(\text{OAc})_2$ , 0.1% BSA, 1 mM DTE, 0.1 mg D-luciferin, 8  $\mu\text{g}$  L-luciferin, 5  $\mu\text{M}$  APS, 0.3 units ATP-sulphurylase, 10 mM  $\text{NaPi}$ , 1  $\mu\text{M}$  DAPP, 10  $\mu\text{g}$  oligomycin, 4  $\mu\text{M}$  antimycin, 5  $\mu\text{M}$  rotenone, purified luciferase (for amount see [23]), chromatophores corresponding to a Bchl concentration of 0.8  $\mu\text{M}$ , 0.5 mM  $\text{NAD}^+$  and 0.5 mM NADPH to start the reaction. Phosphate solutions were incubated with yeast PPase to remove endogenous  $\text{PP}_i$  (for details see [23]).

### 2.3. Continuous monitoring of ATP synthesis

The assay was performed as for  $\text{PP}_i$  determination with the exception that the standard assay contained 50  $\mu\text{M}$  ADP. Oligomycin, APS and ATP-sulphurylase were omitted.

## 3. RESULTS AND DISCUSSION

### 3.1. Measurement and characteristics of the transhydrogenase-driven $\text{PP}_i$ synthesis reaction

Table 1 lists the conditions needed to observe the

Table 1

Reverse transhydrogenase-driven synthesis of  $\text{PP}_i$  and its inhibition by various agents

Conditions	$\text{PP}_i$ synthesis (nmol $\text{PP}_i$ /min per $\mu\text{mol}$ Bchl)
(1) No antimycin and rotenone	340
(2) With antimycin (4 $\mu\text{M}$ ) and rotenone (5 $\mu\text{M}$ )	50
(a) – 10 mM $\text{NaPi}$	0
(b) + 10 mM NaF	0
(c) + 0.1 mM DCCD	0
(d) + 5 $\mu\text{M}$ palmitoyl-CoA	5
(e) + 0.5 mM NADH	0

Continuous  $\text{PP}_i$  synthesis was measured as described in section 2. Variations were omission of antimycin and rotenone (1), omission of  $\text{NaPi}$  (2a), addition of NaF 1–2 min after the start of the reaction (2b), preincubation with DCCD or palmitoyl-CoA for 5 min prior to addition of NADPH (2c,2d) and replacement of NADPH by NADH (2e). The data given have been corrected for  $\text{PP}_i$  synthesis in the dark (see text and fig.1)

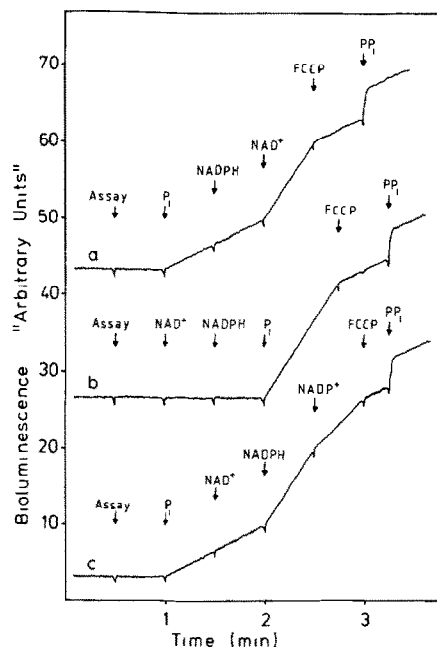


Fig.1. Profile of the reverse transhydrogenase-driven  $PP_i$  synthesis under different conditions.  $PP_i$  synthesis was measured under the conditions described in section 2 and table 1. The traces show the effect on the  $PP_i$  synthesis of (a) reversing the order of addition of  $NAD^+$  and  $NADPH$ , (b) omitting  $P_i$  and (c) adding  $NADP^+$  (0.4 mM). In all cases FCCP was added immediately before the  $PP_i$  standard. Note that the dark reaction  $PP_i$  synthesis is approx. 50% of the reverse transhydrogenase-driven reaction (a,c). The amounts of Bchl and  $PP_i$  standard added were 800 and 10 pmol, respectively.

reverse transhydrogenase-driven  $PP_i$  synthesis and the ways in which it can subsequently be inhibited. In fig.1 we show the time course of  $PP_i$  synthesis under various conditions. To eliminate ATP synthesis or hydrolysis by the  $H^+$ -ATPase, oligomycin was included in the assay system. Similarly, DAPP was included to avoid ATP synthesis by adenylate kinase. Any variations in luciferase and ATP-sulphurylase activity due to sample composition or reagent inactivation were compensated for by internal calibration with a  $PP_i$  standard in every assay.

To observe the true value of the reverse transhydrogenase-driven synthesis of  $PP_i$ , the electron transport inhibitors antimycin A and rotenone have to be included in the assay system (table 1). In the absence of both inhibitors,  $PP_i$  synthesis

was 340 nmol  $PP_i$ /min per  $\mu$ mol Bchl. In the presence of antimycin this was inhibited 35% and in the presence of rotenone alone or of both inhibitors, by 85%. A possible source of energy for the  $PP_i$  produced might be the  $NADH$  generated in the transhydrogenase reaction, which subsequently is oxidized through the rotenone-sensitive  $NADH$ -ubiquinone reductase complex. However, addition of  $NADH$  alone in the presence of antimycin and rotenone produced no  $PP_i$  synthesis (table 1). Therefore, when these inhibitors are present to eliminate this oxidation, the true reverse transhydrogenase-driven  $PP_i$  synthesis is observed, i.e. 50 nmol  $PP_i$ /min per  $\mu$ mol Bchl. This value is approx. 50% of the light-driven synthesis under the same conditions (cf. [22]). The order of addition of  $NADPH$  or  $NAD^+$  has no effect on the profile of the reverse transhydrogenase-driven  $PP_i$  synthesis (see fig.1). Confirmation that the  $PP_i$  synthesis is catalysed by the  $H^+$ -PPase is provided by absence of  $PP_i$  synthesis when the substrate  $P_i$  is omitted or when the enzyme is specifically inhibited using NaF (table 1). Similarly, blocking the proton channel of both enzymes using DCCD also leads to inhibition of reverse transhydrogenase-driven  $PP_i$  synthesis (table 1). In addition to blocking proton channels [27], DCCD has also been reported to inhibit the catalytic activity of the transhydrogenase [28].

The results in figs 1 and 2a also provide direct evidence that the driving force for this  $PP_i$  synthesis by the  $H^+$ -PPase is the  $\Delta\mu H^+$  generated by the reverse transhydrogenase reaction.  $PP_i$  synthesis is inhibited by preincubation with the specific transhydrogenase inhibitor palmitoyl-CoA (table 1). Rydström [29] first showed that palmitoyl-CoA inhibits both the forward and reverse transhydrogenase reactions in submitochondrial particles. Our observation that 5  $\mu$ M palmitoyl-CoA causes 90% inhibition of  $PP_i$  synthesis is in good agreement with his results [30]. Also, the rate of  $PP_i$  synthesis is inhibited by  $NADP^+$ , one of the products of the reaction, with 50% inhibition at 0.4 mM  $NADP^+$  (figs 1c, 2a).

The  $PP_i$  synthesis observed prior to initiation of the reverse transhydrogenase-driven  $PP_i$  synthesis (typically 50%, see fig.1) is assumed to be a dark reaction which bypasses the requirement for  $\Delta\mu H^+$  and is due to the high  $\Delta G^\circ$  of  $-11.4$  kcal/mol for the ATP-sulphurylase-catalysed reaction [31],

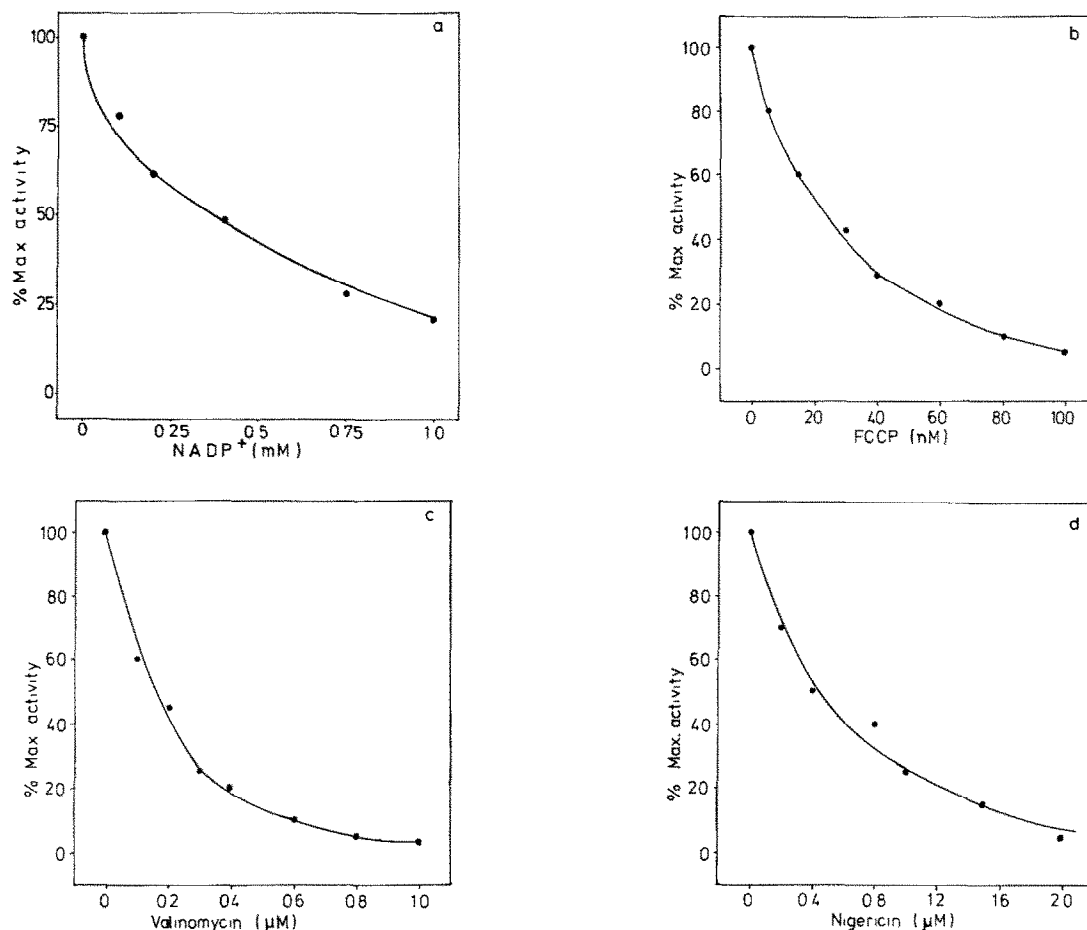


Fig.2. Titration of the reverse transhydrogenase-driven  $PP_i$  synthesis with product uncoupler and ionophores.  $PP_i$  synthesis was titrated with (a)  $NADP^+$ , (b) FCCP, (c) valinomycin and (d) nigericin. 100% activity values were measured as described in section 2 and have been corrected for dark reaction  $PP_i$  synthesis. Valinomycin and nigericin were incubated for 5 min prior to the addition of NADPH.  $NADP^+$  and FCCP were added after the reverse transhydrogenase-driven  $PP_i$  synthesis had been established. The amounts of Bchl and  $PP_i$  added were as in fig.1.

which drives the  $H^+$ -PPase-catalysed reaction ( $\Delta G'_0 = -4$  kcal/mol) [32]. This is supported by the observation that inhibition of the  $H^+$ -PPase (by NaF) causes inhibition of this dark reaction.

### 3.2. Dissipation of the $\Delta\mu H^+$ set up by the reverse transhydrogenase reaction

Fig.2b-d shows the results of titrating the reverse transhydrogenase-driven  $PP_i$  synthesis in chromatophores with the uncoupler FCCP and with the ionophores valinomycin and nigericin, the last two in the presence of 20 mM KCl. 50% inhibition was obtained using 20 nM FCCP, 0.15  $\mu$ M valinomycin and 0.4  $\mu$ M nigericin. It should be noted that the

capacity to form a  $\Delta\mu H^+$  is low in this system under these conditions when compared to the conditions which lead to light-induced  $PP_i$  synthesis [22].

### 3.3. ATP synthesis driven by the reverse transhydrogenase reaction

In fig.3 is shown the time course of reverse transhydrogenase-driven ATP synthesis during which the rate reached 35 nmol/min per  $\mu$ mol Bchl. This is significantly lower than the rate of  $PP_i$  synthesis observed. In this experiment the  $\Delta\mu H^+$  generated is being used both for  $PP_i$  synthesis by the  $H^+$ -PPase (although this  $PP_i$  is not

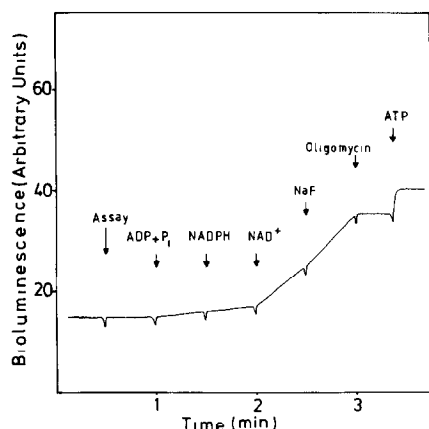


Fig.3. Profile of the reverse transhydrogenase-driven ATP synthesis. ATP synthesis was measured under conditions described in section 2. The trace shows the effect of NaF and oligomycin on ATP synthesis. The amounts of Bchl and ATP standard added were 800 and 10 pmol, respectively. Note that the response of the luciferase system is inhibited by 20% after addition of 10 mM NaF. This effect is not shown in the trace.

detected because there is no ATP-sulphurylase and APS present), and for ATP synthesis by the  $H^+$ -ATPase. Addition of NaF inhibits the  $H^+$ -PPase and so the  $\Delta\mu H^+$  is used to a greater extent for ATP synthesis, hence the increase of 30% in synthesis of ATP. Consequently, addition of oligomycin leads to complete inhibition of ATP synthesis.

In conclusion, we have shown that it is possible to reverse the energy-linked transhydrogenase enzyme of *R. rubrum* chromatophores and use the  $\Delta\mu H^+$  generated to drive  $PP_i$  synthesis by the  $H^+$ -PPase and ATP synthesis by the  $H^+$ -ATPase.

It is worthwhile to note that at the low energy levels obtainable by the reversal of the transhydrogenase reaction, the rate of ATP synthesis is only 70% of the rate of  $PP_i$  synthesis. This is in striking contrast to the situation when saturating light is the energy source. The rate of ATP synthesis is then 7–10-times greater than that of  $PP_i$  synthesis. On the other hand, at very low light intensity, or after single light flashes, where the energy levels also are low, the  $PP_i$  synthesis is significantly higher than the ATP synthesis [22,33].

#### ACKNOWLEDGEMENTS

This work was supported by a grant from the

Swedish Natural Science Research Council to M.B. and by an Imperial Chemical Industries (UK) postdoctoral fellowship to I.H. We would like to thank Dr J. Rydström for valuable discussion.

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